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Synthesis and SAR of 2H-1,2,4-Benzothiadiazine-1,1-dioxide-3-carboxylic acid Derivatives as Novel Potent Glycine Antagonists of the NMDA Receptor-Channel Complex

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Abstract: A new series of 2H-1,2,4-benzothiadiazine-1,1-dioxide-3-carboxylic acids was synthesized. Most compounds demonstrated excellent binding affinities at the glycine binding site of the NMDA receptor-channel complex. A member of this series, **RPR 104632** displaces strychnine-insensitive [3H]-DCKA binding with high potency ($^1C_{50} = 8$ nM) and stereoselectivity: the levorotatory isomer (-)-8k ($^1C_{50} = 4$ nM) is about 500-times more potent than the dextrorotatory isomer (+)-8k ($^1C_{50} = 1880$ nM). In <u>in vitro</u> and <u>in vivo</u> models, **RPR 104632** is a potent antagonist of central NMDA receptor activation.

L-glutamic acid is the major endogeneous excitatory amino acid (EAA) neurotransmitter in the mammalian central nervous system. There is considerable evidence that the overstimulation of excitatory amino acid receptors, notably the NMDA (N-methyl-D-aspartic acid) and AMPA [2-amino-3-(3-hydroxy-5-methylisoxazole-4-yl)propionic acid] receptor sub-types, may be associated with various acute and chronic neurodegenerative disorders.¹

Johnson and Ascher (1987) demonstrated that glycine acts as a coagonist at a specific site on the NMDA receptor² to amplify the agonist action of glutamic acid. This mechanism is independent of its known action on the strychnine-sensitive glycine receptor. These findings have led to an intensive search for NMDA antagonists that inhibit this action of glycine. Several compounds belonging to various chemical families have been shown to be effective glycine-site NMDA antagonists.³

$$R_{2}$$
 R_{2}
 R_{1}
 R_{2}
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 R_{2}
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 R_{5}
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[#] Dedicaced to Prof. H. G. VIEHE on the occasion of his 65th birthday

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7	R ₁	IC ₅₀ a	8c	R ₄	IC ₅₀ ª	8c	R ₄	IC ₅₀ a
7a	-CH₃	400	8a	2-CH ₃	80	8m	3-CN	17
7b	-C ₆ H ₅	2600	8b	2-OCH ₃	21	8n	3-NO ₂	12
7c	-CH ₂ -C ₆ H ₅	48	8c	2-CF ₃	100	80	3-OCF ₃	20
7d	-(CH ₂) ₂ -C ₆ H ₅	90	8d	2-F	26	8р	3-OC ₆ H ₅	9
78	-(CH ₂) ₃ -C ₆ H ₅	33	8 e	2-Cl	48	8q	3-NH ₂	29
7f	-(CH ₂) ₄ -C ₆ H ₅	77	8f	3-CH ₃	14	8r	3-C ₆ H ₅	20
7g	-(CH ₂) ₅ -C ₆ H ₅	36	8g	3-OCH ₃	12	8s	4-CH ₃	45
7h	-CH ₂ -CO ₂ H	9500	8h	3-CF ₃	18	8t	4-0CH ₃	40
7i	-(CH ₂) ₂ -N(CH ₃) ₂	3600	81	3-F	13	8u	4-CF ₃	92
7 j	-CH ₂ -CO-C ₆ H ₅	22	8j	3-CI	11	8v	4-F	44
7k	-(CH ₂) ₂ OC ₆ H ₅	41	8k ^b	3-Br	8	8w	4-Ci	20
71	-(CH ₂) ₂ SC ₆ H ₅	38	(-)-8k	3-Br	4	DCKA		46
7m	-(CH ₂) ₂ SO ₂ C ₆ H ₅	70	(+)-8k	3-Br	1880	2		50
			81	3-1	8	L-689,560		26

Table 1: In Vitro Activities of 2, 7a-m, 8a-w, DCKA, and L-689,560

^a: IC_{50} values (nM) are the mean of the last 3 determinations each with 6 concentrations of test compounds triplate. ^b: RPR 104632 ^c: $R_1 = CH_2(R_4-Ph)$

Scheme 1; Synthesis of Compounds 7a-I and 8 a-p, r-w.

CI
SO₂CI
NH₂
10
Pathway A

CI
SO₂NHR₁

Pathway B

CI
SO₂NHR₁

CI
SO₂NHR₁

OI
NH₂
11

Pathway B

CI
SO₂NHR₁

OI
NH₂
13

CI
SO₂NHR₁

OI
NH₂
14

OI
NH₂
15

Pathway B: 7a, I-I and 8a, d-h, jk, mn, p, r-t, vw

Reagents and reaction conditions: a) CISO $_3$ H, 105°C, 24 h, 84% crude b) liq. NH $_3$, 1.5 h, 73% c) R $_1$ -NH $_2$. THF, (C $_2$ H $_5$) $_3$ N, rt, 10-24h d) OHC-CO $_2$ H . H $_2$ O, C $_2$ H $_5$ OH, H $_2$ SO $_4$, reflux, 3 h then LiOH, THF-water mixture, rt, 12 h for compounds 7b and 8u or (CH $_3$) $_3$ SiOK, CH $_2$ CI $_2$, rt, 24 h for compound 7c or then 0.5N NaOH. rt, 12 h for compounds 7d and 8bl; OHC-CO $_2$ H . H $_2$ O, 1.4-dioxane-water mixture, reflux, 2-8 h for compounds 7e-h and 8c,l,o e) OHC-CO $_2$ H . H $_2$ O, C $_2$ H $_5$ OH, H $_2$ SO $_4$, reflux, 1 h f) NaH, THF, rt then R $_1$ -X (X = Cl or Br) in THF, reflux 2-5 h g) (CH $_3$) $_3$ SiOK, CH $_2$ CI $_2$, rt, 1-24 h for compounds 7a,k; LiOH, THF-water mixture, rt, 24 h for compounds 8a,d-h,j,s,t,v,w; 0.5N NaOH, rt, 24 h for compounds 7i,J,I and 8k,m,n,p,r.

A number of these agents, such as kynurenic acid derivatives (e.g. 5,7-dichlorokynurenic acid, **DCKA**), 2-carboxy-indole-3-propionic acid derivatives (e.g. 2) and 5,7-dichloro-4-ureido-2-carboxytetrahydroquinoline derivatives (e.g. **L-689,560**) have been reported 3 to be antagonists of the NMDA receptor-channel complex. In our assays, 4 these compounds display IC₅₀ values for the glycine modulatory site of 46, 50 and 26 nM, respectively (Table 1).

This paper reports the synthesis and SAR of a new class of potent and selective glycine-NMDA antagonists **7a-m** and **8a-w** (Table 1 and Scheme 1). One of them, compound **8k** (**RPR 104632**) displays excellent affinities for the glycine binding site, and possesses potent antagonist properties in functional models. The biological activity is restricted to the levorotatory isomer (-)-**8k**.

We initially performed the synthesis of the quinazoline-2-carboxylic acid derivative 4, which possesses the general structure of kynurenic acid 1, with an additional nitrogen atom in the 3-position. The affinity for the glycine site decreased dramatically with respect to compound 1 ($IC_{50} > 100 \mu M \ vs. 16 \mu M$). Next, we replaced the carboxamido group of 4 by the corresponding sulfonamido moiety to give the benzothiadiazine 5. Unexpectedly,

this compound displayed a 10 μ M affinity for the glycine site, close to that of kynurenic acid 1. Starting from this interesting result, we synthesized 3,4-dihydro derivatives of these 2H-1,2,4-benzothiadiazine-1,1-dioxide-3-carboxylic acids with the general formula 6. The parent compound 6a (R₁,R₂,R₃=H) attained a similar affinity for the glycine

binding site with an IC50 value of 8 µM.

It is well known that in the kynurenic acid, tetrahydroquinoline acid or 2-carboxy-indole acid series, affinity is greatly enhanced by appropriate introduction of chlorine atom(s) to the phenyl ring in position(s) 5, and 7, 5 and 7 or 4 and 6, respectively.³ We observed the same effect in the 3,4-dihydro-1,2,4-benzothiadiazine series. Thus, the compounds 6a-h 5 displayed affinities for the glycine site with the following rank order of binding: 6b (R₁ = H; R₂,R₃ = 6,8-Cl; 0.073µM) > 6c (R₂ = 6-Cl; R₁,R₃ = H; 0.18µM) > 6d (R₁ = H; R₂,R₃ = 6,8-CH₃; 0.32µM) > 6e (R₂ = 6-CH₃; R₁,R₃ = H; 0.41µM) > 6f (R₂ = 8-l; R₁,R₃ = H; 0.79µM) > 6g (R₁ = H; R₂,R₃ = 6,7-Cl; 1.9µM) > 6a (R₁,R₂,R₃ = H; 8.5µM) > 6h (R₁ = H; R₂,R₃ = 7,8-Cl; 27µM).

Starting from these results, we prepared the compounds **7a-m**⁵ and the benzyl derivative analogs **8a-w** ⁵, varying the substituents in position 2 (Table 1).

Synthesis: (Scheme 1) Compounds **7a-I** and **8a-p,r-w** were synthesized by two approaches. The first approach (**pathway A**) was the reaction of the aminophenylsulphonyl chloride **10** ⁶ with the amine **12** in moderate to high yields (14 - 90%) to give sulphonamides **13**. The desired compounds **7b-h** and **8b,c,I,I,o,u** were obtained in

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moderate to high yields (17 - 87%) from the corresponding sulphonamides 13, by reaction with glyoxylic acid. The second approach (pathway B) required the addition of glyoxylic acid in acidic medium in the presence of ethanol to the sulphonamide 11 ⁶ to provide 14 in 86% yield. Then, the alkylation of the anion formed from 14 with the alkyl halides 15 afforded the corresponding ethyl-2-alkylbenzothiadiazinecarboxylate derivatives 16 in moderate to good yields (22 -61.5%). Aqueous basic hydrolysis of 16 gave 7a,I-I and 8a,d-h,Jk,mn,p,r-t,vw in 25 - 94% yield.

The compound 7m was synthesized by oxidation of 7l (mCPBA, CH_2CI_2 , rt, 0.5 h) in 60 % yield whereas the compound 8q was obtained by reduction of the corresponding nitro derivative 16 ($R_4 = NO_2$) under hydrogen pressure (1 bar, 5% Pd/C, ethanol, rt, 12 h) followed by an aqueous basic hydrolysis (0.5N NaOH) in 56% overall yield.

All new compounds have been caracterized by ¹H-NMR, IR and Mass Spectroscopy, and have given satisfactory combustion analyses (C, H, N, O, S). ⁷

Biological Activity:

The IC_{50} values ⁴ for compounds **2**, **7a-m**, **8a-w**, **DCKA**, and **L-689,560** are shown in Table 1. Among the compounds prepared, the 3-bromobenzyl-derivative **8k** most potently displaced [3 H]-DCKA binding (1 C₅₀ = 8 nM). On the basis of the binding data, the following structure-activity relationships were observed:

• Introduction of an alkyl group in the 2-position: a methyl group reduced the binding affinity (7a vs. 6b) while compound 7b bearing a phenyl ring is much less active ($IC_{50} = 2.6 \,\mu\text{M}$) • Introduction of phenylalkyl chains retained potency whatever the chain length (7c-g vs. 6b) • Introduction of an alkyl chain bearing either a carboxylic acid or an amino function dramatically reduced potency (7h and 7i vs. 6b). • Introduction of either an oxygen, sulfur atom or a sulfoxide moiety onto the phenylalkyl chain of compound 7d had little effect on the affinity for the glycine site (7k-m vs. 6b).

These results suggest the presence of a large bulk tolerance in the vicinity of the 2-substituent and a variety of groups are tolerated without loss of affinity. The excellent binding affinities of phenylalkyl-substituted benzothiazoline $\mathbf{7c}$ ($\mathbf{1C_{50}} = 48 \text{ nM}$) led us to choose the benzyl group for framework optimization. Consequently, a series of 2-benzylsubstituted compounds $\mathbf{8a-w}$ were prepared (Table 1). Activity is optimal when there is a substituent in the 3-position of the phenyl ring compared to substitution in the 2- or 4-positions ($\mathbf{8f-j}$ vs. $\mathbf{8a-e}$ or $\mathbf{8s-w}$).

Derivatives bearing various 3-substituents such as electron-withdrawing groups (e.g. -CN, -NO₂), electron-donating groups (e.g. -OCH₃, -OC₆H₅, -NH₂) or halogen atoms are 2 to 5-fold more active than the corresponding parent compound **7c**. These results suggest that the increase in affinity of 3-substituted-benzyl derivatives relative to **7c** may be a consequence of enhanced hydrogen bonding ability of these polar groups.

As shown in Table 1, RPR 104632 displays very high affinity for the [3 H]-DCKA binding site (1 C₅₀ = 8 nM), six-and three-fold more potent than DCKA and L-689,560 respectively.

Given the high glycine-binding site affinity of racemic RPR 104632, we went on to examine both enantiomers (+)-8k and (-)-8k of this compound. The two enantiomers were prepared in optically pure form by HPLC using a column packed with a chiral stationary phase (Pirkle-type phase Chyrosine A from SEDERE-France). Enantiomeric homogeneity of both enantiomers (>99%) was evaluated by HPLC using the same chiral phase (Table 2).

Table 2: Chromatographic separation of enantiomers (+)-8k and (-)-8k

	retention time	weight	yield	$[\alpha]_{D}^{20}$ (c = 0.5; MeOH)
(+)-8k	24 mn	0.335 g	92%	+ 32.6 ± 1°
(-)-8k	30 mn	0.300 g	82%	- 30.7 ± 1.2°

As shown in Table 1, the levorotatory isomer (-)-8k displayed about a 500-fold greater potency at the glycine site than the dextrorotatory (+)-8k ($IC_{50} = 4 \text{ vs.}$ 1880 nM, respectively).

RPR 104632 showed potent and selective antagonist activity in several functional models. Thus, RPR 104632 antagonized concentration-dependently ($IC_{50} = 890$ nM) the rise in cGMP levels induced by NMDA (80μ M) stimulation ⁸ in immature rat cerebellar cortex, an effect that was reversed by addition of 1 mM glycine. Intravenous administration of 1-30 mg/kg RPR 104632 reduced microiontophoretic responses of thalamic neurons to NMDA .⁹ The maximal decrease reached 70 % inhibition and lasted for 60 min. *In vivo*, RPR 104632, injected i.c.v., dose-dependently protected DBA/2 mice from audiogenic seizures ¹⁰ with an ED₅₀ of 2.5 μ g/mouse. However, RPR 104632 was inactive when injected systemically at 100 mg/kg i.p.

RPR 104632 proved to be highly selective for the glycine binding site. Considerable crossover of activity between the glycine-site and the AMPA receptor has been observed for previously-described series of antagonists. RPR 104632 displaced [3 H]AMPA binding with an IC $_{50}$ of 38 μ M and is thus ~5000-fold selective for the glycine-site over the AMPA receptor. No displacement of radioligands for α_1 -, α_2 - or β -adrenergic, muscarinic, D_1 - or D_2 -dopamine, 5-HT $_{2A}$ -serotonin or Ca $^{2+}$ channel receptors was observed at concentrations of 10 μ M.

In conclusion, RPR 104632 is representative of a novel chemical series and is easily prepared in a five-step synthesis from commercially available reagents. In *in vitro* and *in vivo* models, RPR 104632 is one of the most potent glycine site antagonist reported to date ¹¹ in measures of stimulation of central NMDA receptors. Further pharmacological studies to evaluate the efficacy of RPR 104632 and its enantiomer (-)-8k as neuroprotective agents are in progress and will be reported elsewhere.

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References and Notes.

- § To whom correspondence should be addressed.
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- 7. Melting point (°C), 7a: 260, white solid; 7b: >250, white solid; 7c: 219, white solid; 7d: 83, white solid; 7e: 290, cream-coloured solid; 7f: 67, cream-coloured solid; 7g: 130, sodium salt, white solid; 7h: 257, white solid; 7l: 224, white solid; 7j: 130, white solid; 7k: 211, white solid; 7l: 224, white solid; 8a: 95, white solid; 8b: 80, white solid; 8c: 129, white solid; 8d: 110, white solid; 8e: 90, cream-coloured solid; 8f: 100, white solid; 8g: 109, white solid; 8h: 98, white solid; 8l: 190, white solid; 8j: 105, white solid, 8k: 217, white solid; 8l: 130, white solid; 8m: 238, white solid; 8n: 100, yellow solid; 8o: 115, white solid, 8p: 80, white solid; 8q: 200, white solid; 8r: 100, white solid; 8s: 103, white solid; 8t: 110, white solid; 8u: 110, white solid; 8v: 94, white solid; 8w: 125, white solid.
- 8. Accumulation of cGMP in cerebellar slices (0.5 X 0.5 mm) from immature (9 to 12 days) rats was measured according to the method of Ferrandelli *et al.* (*Brain Res.*, **1975**, <u>84</u>, 63).
- 9. Electrophysiological studies: NMDA was applied by microiontophoresis during extracellular unitary recording of thalamic neurons in vivo. Excitatory responses were quantified by measuring every 5 min the amplitude of the integrated response to NMDA. Drug effects after i.v. injection of RPR 104632 were normalized with respect to the pre-drug baseline response.
- 10. Anticonvulsant activity: groups of 6 mice (DBA/2; three weeks old) received i c.v. injection (4μl) of either RPR 104632 or vehicle by i.c.v. route (4 μl). After 15 min, mice were subjected to a high frequency noise (126 db, 12 kHz) for 60 s, which induced in control mice wild running and seizures. ED₅₀ was defined as the dose of drug which totally protected 50% of the mice.
- 11. Recently, Merck Laboratories described 3'-(aryloxy)-3-phenyl-4-hydroxyquinolin-2(1H)-ones as potent orally active antagonists for the glycine binding site (IC₅₀ [³H]-L-689,560 = 1.4 170 nM); Kulagowski *et al. J. Med. Chem.*, 1994, 37, 1402.